Can hydrogen really propel the decarbonization of transportation?

ENST 250: Understanding Energy Professor Brian Spector

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Overview

- Introduction: History of Hydrogen
- Hydrogen Spectrum
- Blue Hydrogen vs Green Hydrogen
- Hydrogen in Transportation
- Conclusion



• Questions and Discussion

History of Hydrogen

- First identified 1776 by Henry Cavendis
- 1804 first internal combustion engine
- Energy storage mention in 1923
- Impractical until nuclear experimentation
- 1970s "Hydrogen Economy"
- Quest for novelty/experimentation

Hydrogen Color Spectrum

- Refers to source/process
- GHG = Greenhouse Gas Footprint
- Grey most common globally
- Black/Brown most common in US

	Terminology	Technology	Feedstock/ Electricity source	GHG footprint*
PRODUCTION VIA ELECTRICITY	Green Hydrogen		Wind Solar Hydro Geothermal Tidal	Minimal
	Purple/Pink Hydrogen	Electrolysis	Nuclear	
			Mixed-origin grid energy	Medium
PRODUCTION VIA FOSSIL FUELS	Blue Hydrogen	Natural gas reforming + CCUS Gasification + CCUS	Natural gas coal	Low
	Turquoise Hydrogen	Pyrolysis	Natural and	Solid carbon (by-product)
	Grey Hydrogen	Natural gas reforming	Natural gas	Medium
	Brown Hydrogen	Gasification	Brown coal (lignite)	High
	Black Hydrogen	Grandrion	Black coal	

*GHG footprint given as a general guide but it is accepted that each category can be higher in some cases.



- Growing industry
- Fertilizer production
- Hydrocracking
- Consumers
- Potential: transportation economy





Hydrogen Spectrum

Grey hydrogen	Blue hydrogen	Green hydrogen
Split natural gas into hydrogen and CO₂	Split natural gas into hydrogen and CO₂	Split water into hydrogen by electrolysis powered by water or wind
CO₂ emitted in the atmosphere	CO₂ stored or reused	No CO₂ emitted

Highest carbon intensity to lowest carbon intensity

Blue Hydrogen

What is blue hydrogen?



Hydrogen's reliance on fossil fuels

In 2019, over 98% of worldwide demand for hydrogen – estimated at 76.5 million metric tons – was supplied by a fossil fuel production method.



Global Hydrogen Production by Method

Blue hydrogen only exists due to CCUS, why not apply that directly to the point of use e.g. power generation?

Figure 1. Blue Hydrogen and End-Use CCUS, from Natural Gas to Heat or Power Generation



 Hydrogen is a prevalent industrial feedstock. Since contemporary, grey, hydrogen production generates substantial CO₂ emissions, blue hydrogen can displace an otherwise high-emitting process.

Steam Methane Reforming (SMR)

- High-temperature steam (700°C–1,000°C) is used to produce hydrogen from a methane source, such as natural gas.
- Reformer: Methane reacts with steam under pressure in the presence of a catalyst to produce hydrogen, carbon monoxide, and a relatively small amount of carbon dioxide.
- Shift Conversion: The carbon monoxide and steam are reacted using a catalyst to produce carbon dioxide and more hydrogen.
- Pressure-swing Adsorption: carbon dioxide and other impurities are removed from the gas stream, leaving essentially pure hydrogen.

Steam-methane reforming reaction

 $CH_4 + H_2O (+ heat) \rightarrow CO + 3H_2$

Water-gas shift reaction

 $CO + H_2O \rightarrow CO_2 + H_2$ (+ small amount of heat)



Global production capacity growth

BLUE HYDROGEN GLOBAL PRODUCTION CAPACITY, ANNOUNCEMENTS BY REGION



GREEN HYDROGEN GLOBAL PRODUCTION CAPACITY, ANNOUNCEMENTS BY REGION



Source: S&P Global Platts Analytics

Blue Project Highlights: UK and Saudi Arabia

- BP announced UK's largest blue hydrogen production facility, H2Teesside, targeting 1GW of hydrogen production by 2030.
- Two million tonnes of CO₂ stored per year → emissions from the heating of one million households.
- Close proximity to North Sea storage sites and existing CO₂ and hydrogen infrastructure→supporting jobs, regeneration and the revitalisation of the surrounding area.
- Industries in Teesside account for over 5% of the UK's industrial emissions and the region is home to five of the country's top 25 emitters.



Saudi Arabia commits \$110B gas field for blue hydrogen development

By MATTHEW MARTIN AND SALMA EL WARDANY on 10/24/2021

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(Bloomberg) - Saudi Arabia said it would use one of the world's biggest natural-gas projects to make blue hydrogen, as the kingdom steps up efforts to export a fuel seen as crucial to the green-energy transition.

A large portion of gas from the \$110 billion Jafurah development will be used for blue hydrogen, according to Energy Minister Abdulaziz bin Salman. It is made by converting natural gas and capturing the carbon dioxide emissions.

"We are the biggest adventurers when it comes to blue hydrogen," Prince Abdulaziz said at a climate conference in Riyadh on Sunday. "We're putting our money where our mouth is on hydrogen. We have a terrific gas base in Jafurah we will use it to generate blue hydrogen."

Caveat: Lifecycle Emissions

- For every unit of heat in the natural gas at the start of the process, only 70-75% of that potential heat remains in the hydrogen product
 - You would need to use 25% more natural gas to make blue hydrogen than if it was used directly for heat.
- US researchers found that methane emissions released when the fossil natural gas is extracted and burned are much less than blue hydrogen.
 - More methane needs to be extracted to make blue hydrogen, and it must pass through reformers, pipelines and ships, providing more opportunities for leaks
 - Research indicates, to make blue hydrogen <u>20% worse</u> for the climate than just using fossil gas.

Cost-benefit Analysis

Strengths

- **Cost-effective**: In 2017, blue hydrogen cost 21.6 cents per mile vs the new gasoline vehicle average at 14 cents per mile
- Scalable: Infrastructure for SMR and hydrogen transportation exists and technical expertise from O&G can be transferred
- Economically opportunistic: Countries with natural gas endowment like US and Saudi are inclined to capitalize on their resources in a sustainable manner

<u>Weaknesses</u>

- Poor Lifecycle Emissions: larger amounts of natural gas required to produce same amount of energy leads to higher methane leak risk
- Reliance on low cost natural gas: Countries without abundant natural gas will have higher costs
- Feasibility of permanent CO₂ storage: Getting permits for permanent storage wells is extremely difficult, and while CCUS is a tested technology, its scale has been relatively small

Green Hydrogen

Green Hydrogen Electrolysis



Source: Solar Thermal World

Green Hydrogen Highlights



Fueling a Toyota hydrogen vehicle in Fountain Valley, Calif. Philip Cheung for The New York Times

Source: NYT

pv magazine

News - Features - Events - Awards - Partner news - pv magazine tes

The Hydrogen Stream: 8 GW green hydrogen project announced in Chile

Source: PV Magazine

SUSTAINABLE ENERGY

Huge \$2.6 billion green hydrogen project planned for Europe

PUBLISHED FRI, DEC 3 2021-7:13 AM EST | UPDATED FRI, DEC 3 2021-7:47 AM EST

Anmar Frangoul





Economic Breakdown of Green Hydrogen Production

Table 1 – Hydrogen costs for PEM electrolysis from H2A with associated inputs of electricity cost, capacity factor, and uninstalled system capital cost.⁴

	Electricity Cost (¢/kWh)	Capacity Factor	System CapEx (\$/kW)	H₂ Cost (\$/kg)
Gridlow	FO	90.0%	1,500	\$5.13
Gria Low	5.0		1,000	\$4.37
Crid Llich	7.0	90.0%	1,500	\$6.27
Gria High			1,000	\$5.50

- Hydrogen can be produced at a cost of ~\$5 to \$6/kg-H2
- Additional costs: electrolyzer (<\$1,500/kW), and grid electricity prices (\$0.05/kWh to \$0.07/kWh).
- PEM: Polymer Electrolyte Membrane electrolyzers
- Grid low/high: low versus high volume electrolyzer capital costs
- Capacity Factor: % of time in operation

Economic Comparison of Green Hydrogen

- LCOH: Levelized Costs of Hydrogen
- Short-term cost: 2.5-6 USD/kg H2
- Long term cost: 1-3.5 USD/kg H2



Future Growth



Analysis Breakdown

Advantages

- Sustainable: Green hydrogen saves
 830 million tons of CO2 emissions
- Versatile: Green hydrogen can be transformed into electricity or synthetic gas to be used for domestic, commercial, industrial or transportation

purposes

Disadvantages

- **High cost of production**: Energy from renewable sources is more expensive to generate, making hydrogen more expensive to obtain
- High energy consumption: The production of green hydrogen in particular requires more energy than other hydrogens
- **Safety issues**: Hydrogen is a highly volatile and flammable. Extensive safety measures are required to prevent leakage and explosions
- **Transport issues**: High pressure environments are needed to compress Hydrogen while transporting



Potential for Hydrogen - Transport Fuel

	the former
HELER / CIC	
1	
- EMA	

Why Transport Fuel?

• Why not wind and solar?

• Potential benefits not just sustainability?





Current Advancements in Flight

• Smaller planes -> Proven Concept





Established Manufacturers

• Airbus is committed

• Parisian Hub among developments





AIRBUS

Implications

• 2.5 percent of global CO2 emissions





Development Challenges

- Storage
- Airport Infrastructure
- Long range flights
- Production at Scale
- Actual Hydrogen Acquisition





Ships in a similar position

• 3 percent of global CO2 emissions





Limitations as Transport Fuel

Limitations of Hydrogen

- Difficult to Store/Transport
 - Compressed gas or Liquid
- Expensive
- Time Consuming to Make
- Causes Embrittlement
- Highly Flammable and Dangerous



Hydrogen Storage Tank

Limitations of Hydrogen in Jet Fuel



- Airbus released statement:
 - "Hydrogen planes won't take off until 2050"
- Low Volumetric Energy Density
 - New model plane needed
- Reduces passenger capacity
- Limits range



Conclusion: Hydrogen's Competition

- BEV's take longer to charge vs a hydrogen refill
- Battery storage capacity and hence mileage is limited



Cars: direct electrification most efficient by far



Notes: To be understood as approximate mean values taking into account different production methods. Hydrogen includes onboard fuel compression. Excluding mechanical losses.

TRANSPORT & Y R @ In ENVIRONMENT @transportenvironment.org Sources: Worldbank (2014), Apostolaki-Iosifidou et al. (2017), Peters et al. (2017), Larmanie et al. (2012), Umweltbundesamt (2019), National Research Council (2013), Ricardo Energy & Environment (2020), DOE (no date), ACEA (2016).

Questions?